

**SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR**  
(AUTONOMOUS)

**B.Tech I Year II Semester Supplementary Examinations December-2025**  
**CHEMISTRY**

(Common to CSE, CSIT, EEE, ECE)

**Time: 3 Hours**

**Max. Marks: 70**

**PART-A**

(Answer all the Questions 10 x 2 = 200 Marks)

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|-----|---|-----|----|----|
| 1 a | Write short notes on Wave-Particle duality of matter  | CO1 | L2 | 2M |
| b   | Give the Significance of $\Psi$ and $\Psi^2$          | CO1 | L1 | 2M |
| c   | Write the importance on band gap in semiconductors?   | CO2 | L1 | 2M |
| d   | Define Nanomaterial?                                  | CO2 | L1 | 2M |
| e   | Define Single electrode potential?                    | CO3 | L1 | 2M |
| f   | Write the applications of Hydrogen Oxygen fuel cells  | CO4 | L1 | 2M |
| g   | Give examples of Biodegradable polymers?              | CO5 | L1 | 2M |
| h   | Define Polymerization?                                | CO5 | L1 | 2M |
| i   | Give any 4 examples of mobile phase                   | CO6 | L1 | 2M |
| j   | Write the working principle of Liquid chromatography? | CO6 | L1 | 2M |

**PART-B**

(Answer all Five Units 5 x 10 = 50 Marks)

**UNIT-I**

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|---|--|-----|----|-----|
| 2 | Illustrate the molecular orbital diagram of $O_2^+$ and $O_2^{2-}$ . Explain its bond order and magnetic property based on MOT theory. | CO1 | L3 | 10M |
|---|--|-----|----|-----|

**OR**

- |   |  |     |    |     |
|---|--|-----|----|-----|
| 3 | Calculate the bond order of $F_2$ & $NO$ molecule and explain the magnetic properties based on MOT theory. | CO1 | L3 | 10M |
|---|--|-----|----|-----|

**UNIT-II**

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|-----|---|-----|----|----|
| 4 a | Explain in detail about principle and classification of semiconducting materials. | CO2 | L2 | 7M |
| b   | Discuss about Type-I and Type-II Superconductors with examples.                   | CO2 | L2 | 3M |

**OR**

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|-----|--|-----|----|----|
| 5 a | Explain the basic principle and Classifications of Super Capacitors. | CO2 | L2 | 6M |
| b   | Outline the important applications of Graphene nanoparticles.        | CO2 | L2 | 4M |

**UNIT-III**

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|---|--|-----|----|-----|
| 6 | Explain construction and working of Daniel cell? Calculate the single electrode potential of zinc in 0.01M $ZnSO_4$ solution at 298.15 K. $\{E^0_{Zn/Zn^{2+}} = -0.756V\}$ | CO3 | L3 | 10M |
|---|--|-----|----|-----|

**OR**

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|-----|---|-----|----|----|
| 7 a | Define Fuel cell? Describe the Construction and Working principle and uses of Polymer electrolyte membrane fuel cell. | CO4 | L2 | 6M |
| b   | Discuss about potentiometric sensors with examples.   | CO4 | L2 | 4M |

**UNIT-IV**

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|-----|---|-----|----|----|
| 8 a | Explain different types of polymerizations with examples in detail.       | CO5 | L2 | 5M |
| b   | Explain about synthesis, properties and applications of Poly Lactic Acid. | CO5 | L2 | 5M |

**OR**

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|-----|---|-----|----|---|
| 9 a | Explain the preparation, properties and uses of Bakelite. | CO5 | L2 | 5 |
| b   | Explain Co-ordination or Ziegler-Natta polymerization.    | CO5 | L2 | 5 |

**UNIT-V**

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|------|--|-----|----|---|
| 10 a | What is meant by Chromatography? Write about principle and instrumentation of HPLC chromatography with neat diagram.           | CO6 | L2 | 5 |
| b    | What is the use of detector in chromatographic technique and what are the different types of detectors used in HPLC technique. | CO6 | L2 | 5 |

**OR**

- |      |   |     |    |   |
|------|---|-----|----|---|
| 11 a | Explain the various possible electronic transitions occurs in a molecule by absorbing the UV-Visible radiation. | CO6 | L2 | 5 |
| b    | Explain in detail about Stretching and bending vibrations.  | CO6 | L2 | 5 |

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